## THE STRUCTURE OF THE CACTUS STEROL MACDOUGALLIN (14 $\alpha$ -METHYL- $\Delta^8$ -CHOLESTENE- $3\beta$ , $6\alpha$ -DIOL)—A NOVEL LINK IN STEROL BIOGENESIS<sup>1</sup> Sir:

The details of the biosynthesis of cholesterol and related sterols through the sequence mevalonate  $\rightarrow$ farnesol  $\rightarrow$  squalene  $\rightarrow$  lanosterol are in general well documented.<sup>2</sup> One of the most interesting aspects involves the terminal stages in the conversion of lanosterol (I) to cholesterol (III), notably the demethylation<sup>3</sup> of lanosterol (I), which is believed to proceed through initial elimination of the  $14\alpha$ -methyl group, followed by oxidative removal of the two methyl groups at C-4. The operation of this path has been deomonstrated by Bloch and collaborators,<sup>4</sup> who noted the intermediacy of 14-norlanosterol (II) in cholesterol biosynthesis in the rat. Further support has been provided by the isolation, both from animals and plants, of 4-monomethyl sterols.<sup>5</sup>

The isolation and structure elucidation of lophenol<sup>5a</sup>  $(4\alpha$ -methyl- $\Delta$ <sup>7</sup>-cholesten- $3\beta$ -ol) from a cactus has prompted us to search for further links in the lanosterol  $(I) \rightarrow$  cholesterol (III) transformation and we have concentrated on the cactus genera Peniocereus and Wilcoxia, both of which are characterized by large tuberous roots-a rare feature among the Cactaceae. The isolation and characterization of peniocerol ( $\Delta^{8}$ cholestene- $3\beta$ ,  $6\alpha$ -diol) (IV) from *Peniocereus fosterianus* Cut. has already been recorded<sup>6</sup> and mass spectral examination of the mother liquors indicated the presence of a higher homolog with an additional methyl group. Larger amounts of this new sterol, now named macdougallin (first isolated by Dr. R. D. H. Murray in our Laboratory), were encountered together with peniocerol (IV) in Peniocereus macdougalli Cut.7 and its structure elucidation is reported herewith.

Chromatographic separation of macdougallin (V) and peniocerol (IV) was virtually impossible, but fractional crystallization of the diacetates proved successful, the purity of the fractions being established by mass spectrometery in view of the great similarity of their respective infrared spectra. Macdougallin diacetate exhibited m.p. 124–126°,  $[\alpha]_D +55.4^\circ$  (all rotations in chloroform) as well as high terminal ultraviolet absorption ( $\epsilon_{210}$  4800,  $\epsilon_{220}$  1600). This latter feature, coupled with the absence of olefinic proton signals in the n.m.r. spectrum, requires the presence of a tetrasubstituted double bond. Alkaline saponification or exposure to lithium aluminum hydride provided the parent sterol, macdougallin (V), m.p. 173–174.5°,  $[\alpha]_D +71.8^\circ$ , elementary analysis<sup>8</sup> and mass spectrometry establishing

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(2) "Ciba Foundation Symposium on the Biosynthesis of Terpenes and Sterols" (G. E. W. Wolstenholme and M. O'Connor, ed.), J. and A. Churchill, Ltd., London, 1959. For other recent reviews see J. W. Cornforth, Pure Appl. Chem., 2, 607 (1961); L. D. Wright, Ann. Rev. Biochem., 30, 525 (1961).
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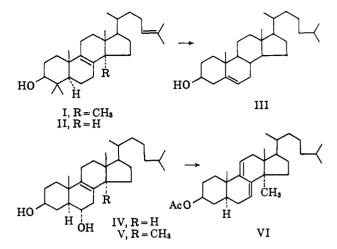
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(6) C. Djerassi, R. D. H. Murray and R. Villotti, Proc. Chem. Soc., 450 (1961).

(7) Collected by Dr. D. K. Cox near La Escondida, Oaxaca (Mexico). We are indebted to Dr. Cox for numerous collections and to Dr. R. Villotti (Syntex, S. A., Mexico City) for the large-scale extractions performed in Mexico.

(8) All subtances reported in this communication gave satisfactory elementary analyses; in most instances the empirical formula was also confirmed by mass spectrometric measurements performed by Drs. M. Ohashi and J. M. Wilson.



the empirical formula  $C_{28}H_{48}O_2$ . In terms of elementary composition, macdougallin (V) thus represents x-methylpeniocerol, a peniocerol (IV) skeleton being assumed because of the close infrared spectral and chromatographic behavior of the two sterols. The *a priori* likely attachment of the extra methyl group at C-4 (as in lophenol<sup>5e</sup>) was excluded by partial saponification of macdougallin diacetate to the 6-monoacetate (amorphous) followed by chromium trioxide oxidation to the 3-keto-6-acetate (m.p. 117–117.5°, positive O.R.D. Cotton effect) and deuterium exchange, the presence of four exchangeable hydrogens being demonstrated by mass spectrometry.

Assuming an intact cholestane skeleton, these results leave only C-14 or the side chain as possible loci for the additional methyl group. An unambiguous answer could not be secured by n.m.r. spectrometry, but the C-14 attachment was made likely by the observation that the nuclear double bond would not be isomerized with acid nor reduced under acidic conditions, in contrast to the behavior<sup>6</sup> of peniocerol (IV). Definite proof for the location of the methyl group could be adduced by treatment of macdougallin diacetate with hydrogen chloride in acetic acid to gave  $14\alpha$ -methyl- $\Delta^{7,9(11)}$ -cholestadien- $3\beta$ -ol acetate (VI), m.p. 77–78°,  $[\alpha]_{D} + 66.3^{\circ}, \epsilon_{234}^{\max} 16500, \epsilon_{243}^{\max} 19400 \text{ and } \epsilon_{251}^{\max} 13900,$ which proved to be identical by mixture melting point determination and infrared comparison with an authentic sample, prepared by selenium dioxide-acetic acid oxidation of synthetic<sup>9</sup>  $14\alpha$ -methyl- $\Delta^7$ -cholesten- $3\beta$ -ol acetate.

Macdougallin is, therefore,  $14\alpha$ -methyl- $\Delta^{8}$ -cholesthene- $3\beta$ , $6\alpha$ -diol (V) and thus represents the first naturally occurring 14-monomethyl sterol. Its existence demonstrates that, at least in the cactus, demethylation in ring A can precede removal of the 14methyl group of a lanosterol-like precursor and, furthermore, raises the question (resolvable by suitable biochemical experimentation) whether this alternate demethylation path may not also be operative in mammalian cholesterol biosynthesis.

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## REACTIVE O-QUINONOID AROMATIC HYDROCARBONS OF THE PLEIADENE SERIES

Sir:

Whereas the known stable hydrocarbons naphthacene and benzo [a] anthracene may be viewed as derived from